

# PID: ToF principle

It requires very good time resolution and a sufficient path L

Given 2 particles  $m_1$  and  $m_2$  with velocities  $\beta_1 c$  and  $\beta_2 c$ :

$$\Delta t = t_1 - t_2 = \frac{L}{\beta_1 c} - \frac{L}{\beta_2 c} = \frac{L}{c} \left( \sqrt{1 + \frac{m_1^2 c^2}{p^2}} - \sqrt{1 + \frac{m_2^2 c^2}{p^2}} \right)$$



Since from  $p = \gamma m v = \gamma m \beta c$  and  $E = \gamma m c^2$

$$\frac{1}{\beta} = \frac{E}{pc} = \frac{\sqrt{p^2 c^2 + m^2 c^4}}{pc} = \sqrt{1 + \frac{m^2 c^2}{p^2}}$$

Given a beam of particles with the same momentum from  $\Delta t$  measurement we separate the 2 species

In our example: given one particle crossing two scintillators if **we know p** and **measure  $\Delta t$  and L** we get m

$$\Delta t = t_1 - t_2 = \frac{L}{\beta c} = \frac{L}{c} \sqrt{1 + \frac{m^2 c^2}{p^2}}$$

# Scintillators in brief

Convert **energy deposited by a charged particles or high energy photons into light**: atoms or molecules of the scintillating medium are excited and decay emitting photons which are detected and converted into electric signals (PMTs)

Scintillating materials:

- **organic**: aromatic hydrocarbon compounds, solid crystals, plastics or liquids, typically they are **faster but have lower light yield**.
- **inorganic**: ionic crystals doped with activator centers or glasses; typically they have **larger light yields**

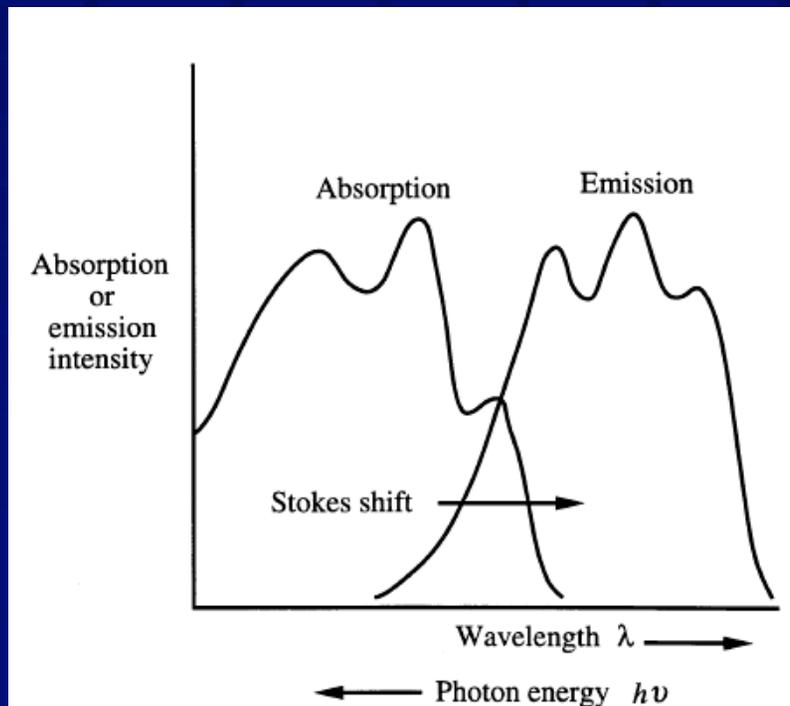
Two types of light emission:

- **Fluorescence**: prompt **ns** → **μs** in visible wavelength range, temperature independent (component useful for particle detection)
- **Phosphorescence**: emission over longer period **μs** → **ms, hrs** with longer wavelength and temperature dependent

# Properties of scintillators

Parameters characterizing scintillators:

- **Efficiency**  $R_s$  = average n. of emitted photons/energy of incident radiation
- **Scintillation yield** =  $R_s h\nu$   $h\nu$  = energy of emitted photons
- **Time response**: depends on decay time of fast component
- the scintillator should be transparent to its own scintillation light (**Stokes' shift**: the emission wavelength is longer than absorption one)
- **linearity**



# Organic scintillators

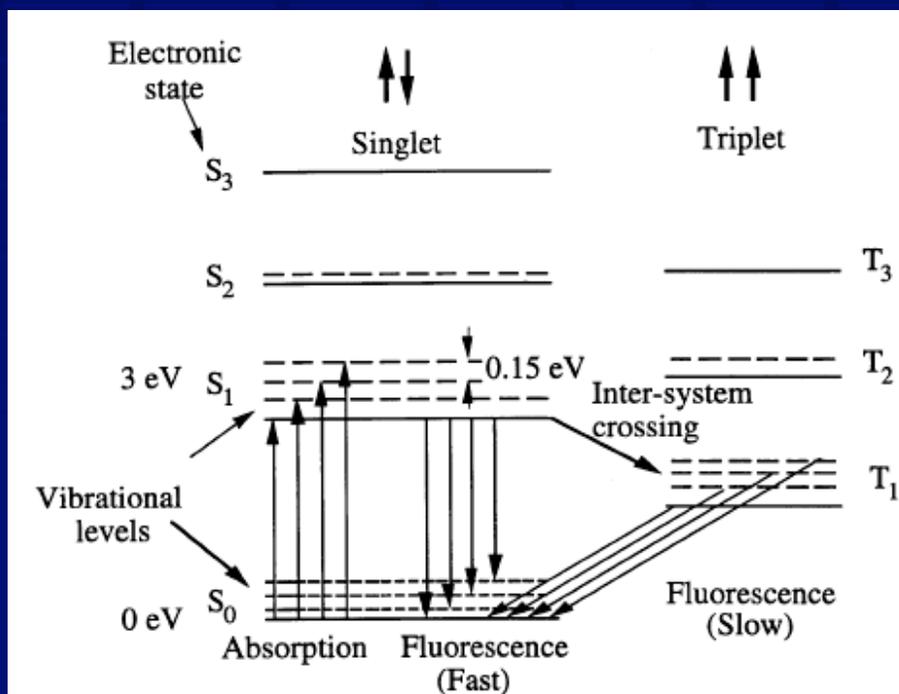
Organic scintillators (plastics or liquids) are composed of aromatic hydrocarbon compounds. Plastic ones are non-fluid solutions consisting of fluorescent organic compounds dissolved in a polymer matrix. Liquid scintillators are fluid solutions with similar fluorescent compounds. No crystal structure is needed.

The emission of light is due to excitation of molecular levels in a primary fluorescent material that emits UV light during de-excitation. This light is absorbed in most organic materials with an absorption length of ~mm. The extraction of a light signal becomes possible only by introducing a second fluorescent material in which the UV light is converted into visible light (wavelength shifter).

Organic scintillators are typically made of low Z materials and have low density. Hence the main interaction >20keV process is not photoelectric absorption (such as in the case of inorganic scintillators) but Compton scattering. Typically, because of the low density more volume is required to obtain a reasonable detection efficiency, but they have low cost.

# Organic scintillators

**Energy levels of organic scintillators:** at room  $T_a$  ( $KT_a = 0.025$  eV) electrons on ground state, incident radiation excites electrons to  $S_1$  states, radiationless decays to base  $S_1$  state, emission of light to  $S_0$ .  $S_1$  can decay to a triplet state with lower energy and longer decay time. The fluorescence UV light (250-370 nm) is absorbed by most organic materials, so the light signal can be extracted using a second fluorescent material (wavelength shifter) that converts UV in visible light (320-500 nm)



# Time dependence of emitted light

Non-radiative transfer of energy from vibrational states to fluorescence state  $S_1$ :  
**0.2-0.4 ns**

Decay of fluorescent state: **1-3 ns**

$$I(t) \propto 1 - e^{-t/\tau_r}$$

$\tau_r$  = rise time constant

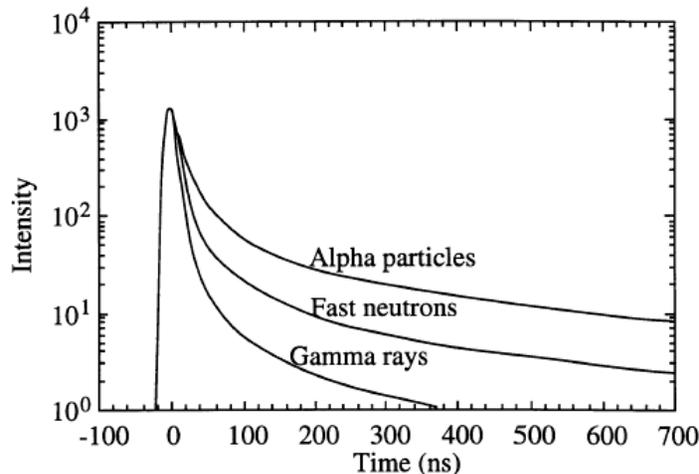
Fall with time constant  $\tau_f$

$$I(t) \propto e^{-t/\tau_f}$$

Material	State	$\lambda_{\max}$ [nm]	$\tau_f$ [ns]	$\rho$ [g/cm <sup>3</sup> ]	photons/MeV
Anthracene	crystal	447	30	1.25	$1.6 \cdot 10^4$
Pilot U	plastic	391	1.4	1.03	$1.0 \cdot 10^4$
NE104	plastic	406	1.8	1.03	$1.0 \cdot 10^4$
NE102	liquid	425	2.6	1.51	$1.2 \cdot 10^4$

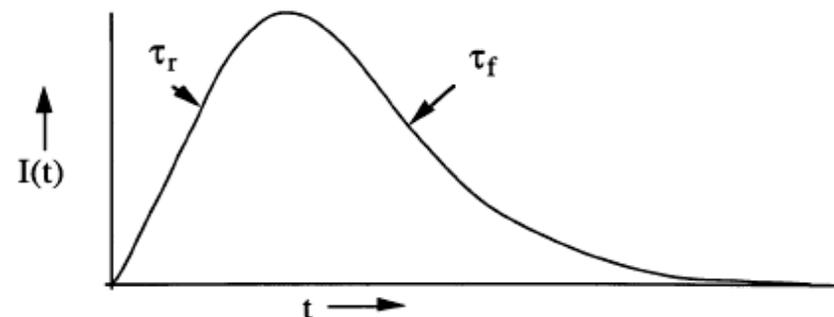
## The decay time depends on the ionization density

Decay time in stilbene for various particles



total pulse shape

$$I(t) = I_0(e^{-t/\tau_f} - e^{-t/\tau_r})$$



# Birk's law

For organic scintillators the relation between emitted light and energy loss is not linear. Deviations from linearity are due to quenching interactions between excited molecules created along the ionizing particle path absorbing energy

For an ideal scintillator and low ionization density

Luminescence  $\propto$  Energy dissipated in scintillator

$$L = SE \quad \text{or} \quad \frac{dL}{dr} = S \frac{dE}{dr}$$

The light output depends on the ionization density

$$\frac{dL}{dr} = \frac{S \frac{dE}{dr}}{1 + kB \frac{dE}{dr}}$$

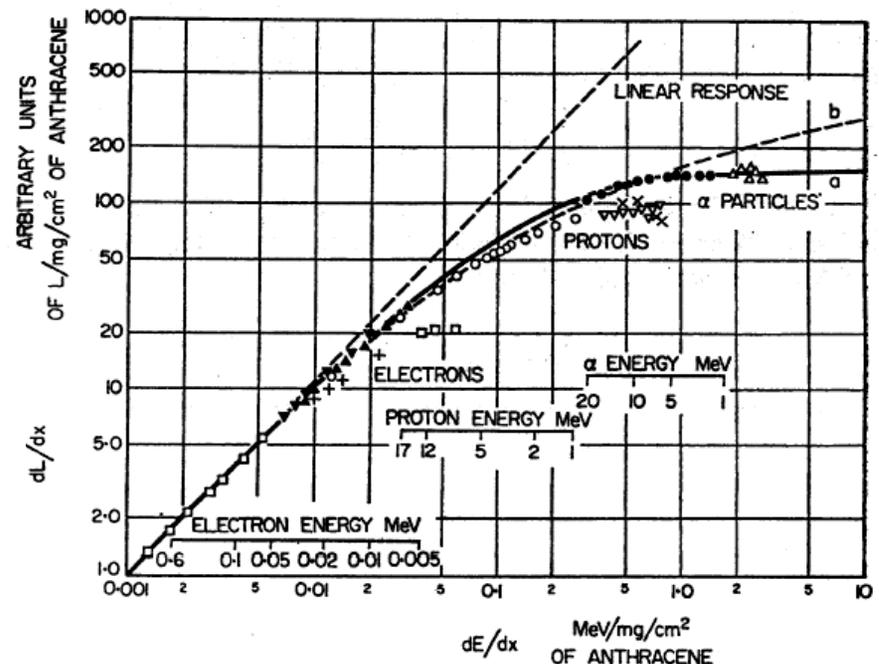
Density of ionized and excited molecules along track

Quenching parameter

For small  $dE/dr$  this yields the luminescence yield postulated above.

For large  $dE/dr$  the specific luminescence saturates, as indicated by the data.

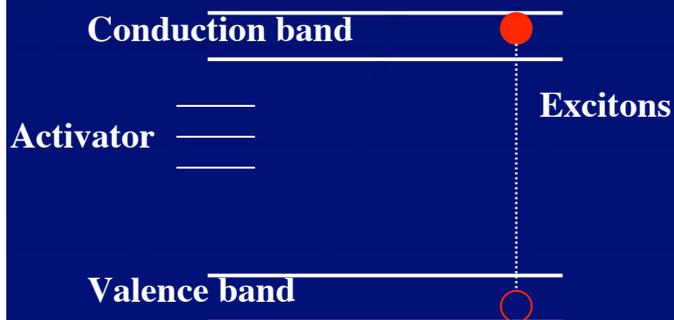
$$\frac{dL}{dr} = \frac{S}{kB} = const$$



# Inorganic scintillators

They are usually made of high Z materials and have a fairly high density. The high Z enhances the **photoelectric interaction** contribution and the high density increases the interaction efficiency. They are **crystals grown in high temperature** furnaces and are made of **Alkali Halides (ie. NaI, CsI) or Oxides (eg. BGO)**. They have scintillation properties by virtue of their crystalline structure that creates the energy bands between which electrons can jump. Some crystals need activators to enable emission in the visible (eg Thallium which is used in the most frequently used inorganic scintillator NaI(Tl)). Ionizing particles produce free electrons, holes and couples of electron-holes (excitons). These move around the crystal lattice until they meet an activator center that they transform into an excited state  $A^*$  of energy  $E_1$  that can decay emitting light. **The decay time depends on the temperature as  $\exp[-E_1/(KT)]$**

Energy level of a ionic crystal doped with an activator



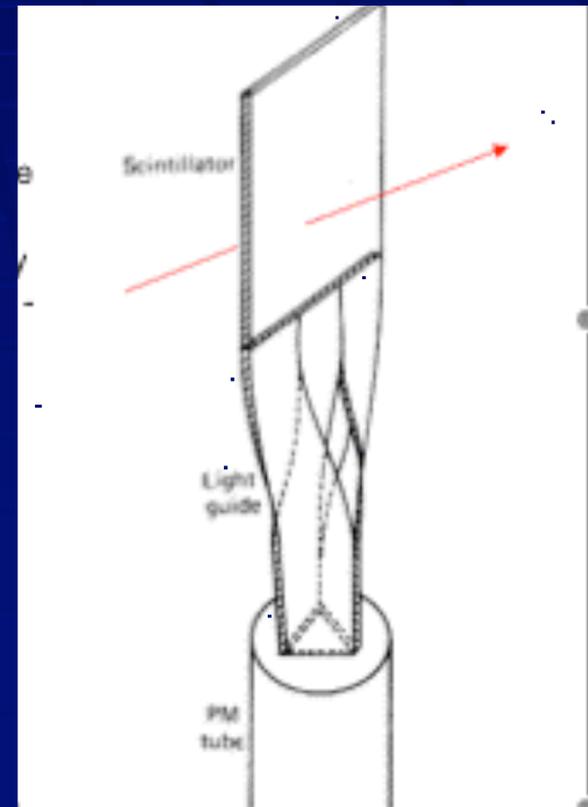
Material	Form	$\lambda_{\max}$ (nm)	$\tau_f$ (ns)	$\rho$ (g/cm <sup>3</sup> )	Photons per MeV
NaI(Tl) (20°C)	crystal	415	230	3.67	38,000
pure NaI (-196°C)	crystal	303	60	3.67	76,000
Bi <sub>4</sub> Ge <sub>3</sub> O <sub>12</sub> (20°C)	crystal	480	300	7.13	8,200
Bi <sub>4</sub> Ge <sub>3</sub> O <sub>12</sub> (-100°C)	crystal	480	2000	7.13	24,000
CsI(Na)	crystal	420	630	4.51	39,000
CsI(Tl)	crystal	540	800	4.51	60,000

# Scintillators coupled to PMTs

A plastic light guide couples the large cross section scintillator to the small area PMT. The light emitted by the scintillator is transported to the photocathode through the light guide through total internal reflections (when the incident angle is larger than the critical one  $\theta_c = \sin^{-1}(n_{\text{med}}/n)$   $n_{\text{med}} = 1$  for air)

Typically fractions of trapped light are around 20%, the scintillator and light guide may be wrapped of Aluminium foil to prevent leaks of light escaping internal reflections. They are wrapped of black layers to prevent outside light to enter.

The first PMT was built in 1913 by Elster and Geiger  
RCA made them commercial in 1936



# Spectral regions and Units

Wavelength (nm)	Spectral range	Frequency (Hz)	Energy (eV)
10	X-ray Soft X-ray		$10^2$
100	Extreme UV	$10^{16}$	10
350	UV	$10^{15}$	
750	Visible		
1000	Near IR		
10000	IR	$10^{13}$	$10^{-1}$
$10^6$	Far IR	$10^{12}$	$10^{-3}$

# Phototubes

[http://sales.hamamatsu.com/assets/applications/ETD/pmt\\_handbook\\_complete.pdf](http://sales.hamamatsu.com/assets/applications/ETD/pmt_handbook_complete.pdf)

Transform a light signal into an electric one: photons hitting the thin photocathode layer of an evacuated glass or quartz tube extract photoelectrons (photoelectric effect)

The energy of the emitted electron is  $E = h\nu - \Phi$ ,  $\nu$  = frequency of photon and  $\Phi$  = work function of the photocathode

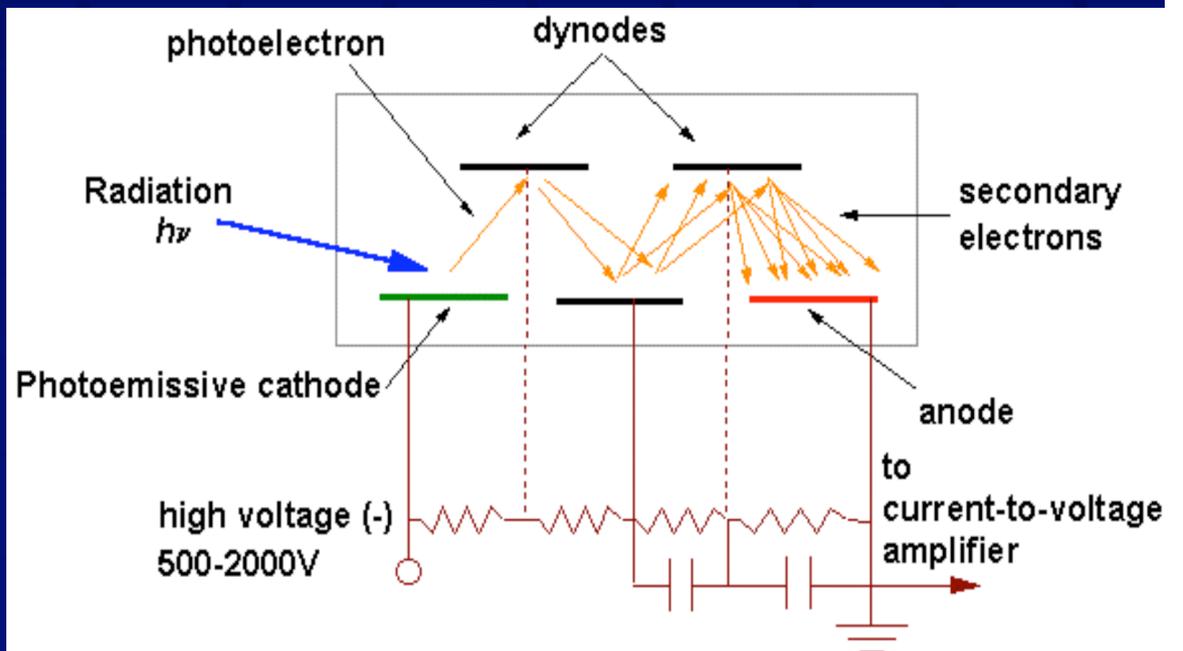
Photocathodes are semiconducting alloys containing 1 or more metals from the alkali group (Na, K, Cs) and materials from group V (eg Sb - antimony).

Bialkali cathodes are made of 2 alkali components.

$QE(\nu) = \#pe \text{ emitted by the photocathode/incident photons} \sim 20\text{-}25\%$  at maximum ( $\lambda \sim 100\text{-}1200 \text{ nm}$ )



ANTARES  
10 inch Hamamtsu 14 stages



# Dynode structures

The collection efficiency of the 1st dynode = # of electrons hitting it/#of emitted photoelectrons ~ 60-90%

Secondary emissive materials used for dynodes are alkali antimonide, Be or Mg oxyde (BeO). Gallium phosphide (GaP) and gallium arsenide phosphid.

$\delta$ =secondary emission ratio=  
#of secondary  $e^-$  per primary  $e^-$   
Varies for various materials  
as a function of the  
accelerating voltage of primary  $e^-$

Venetian blind

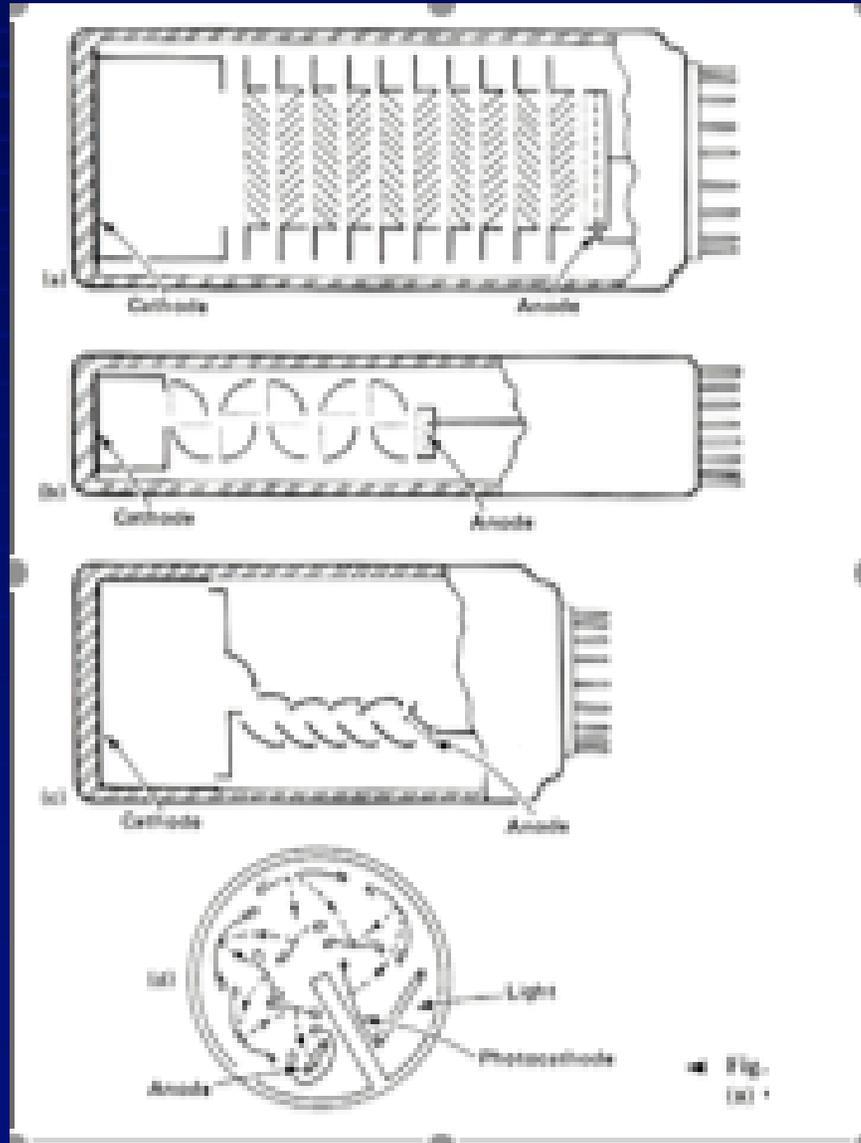
Box and grid

Linear-focused

Circular cage



Secondary emissive surface



# Photosensors



IceCube

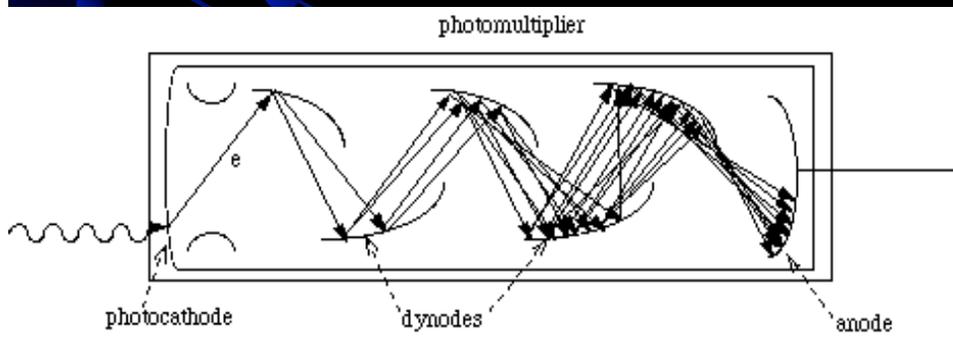
10 inch Hamatsu 10 stages

Gain  $\propto V^{KN}$   $N = \#$  dynodes  
 $K = 0.7-0.8$  depends on  
structure and material of  
electrodes

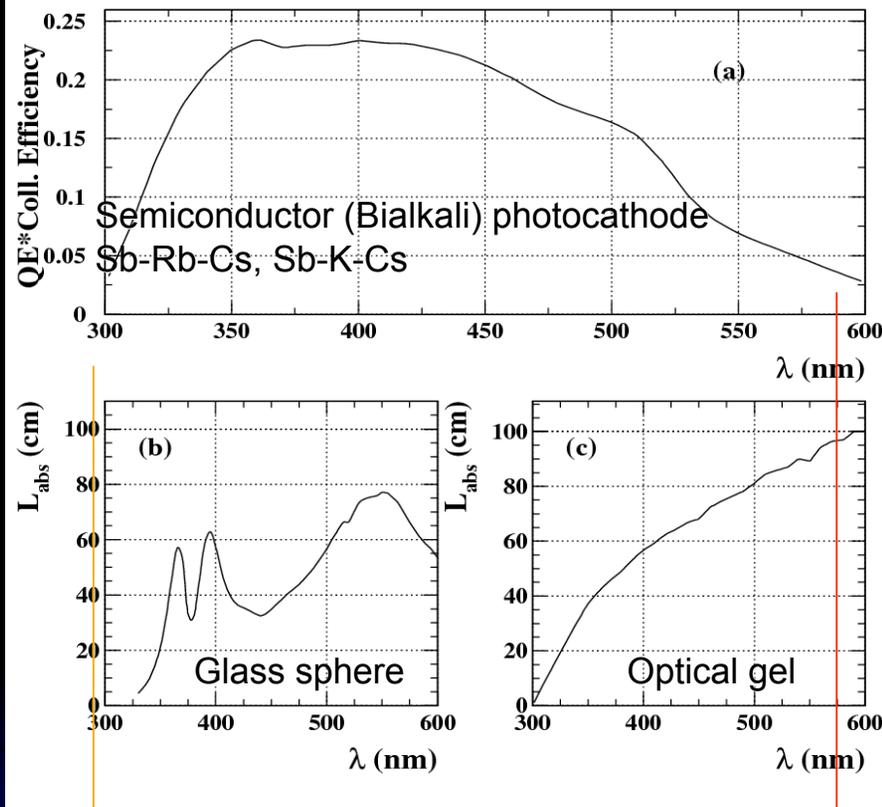
pe's undergo an amplification hitting the dynodes  
Dynodes are made of materials with high  
secondary electron emission (BeO, Mg-O-Cs) and  
are kept at potential differences (150-200 V) that  
accelerate electrons. Multiplication of the number  
of electrons can reach up to  $10^8$ .

This charge of  $10^8 \times 1.6 \cdot 10^{-19} \text{ C} \sim 2 \cdot 10^{-11} \text{ C}$  arrives  
at the anode in a time interval  $\sim 5 \text{ ns} \Rightarrow I = 4 \text{ mA}$ . If  
the anode is connected to ground via a  $R = 50 \text{ W}$   
 $\Rightarrow$  a pulse of 200 mV is formed. While the rise time  
of the pulse is  $\sim 2 \text{ ns}$  the transit time in the tube is

$\sim 40 \text{ ns}$  and the Transit Time  
Spread (spread in times of  
propagation of electrons from  
the photocathode to the first  
dynode due to the distribution of  
the velocity of emitted  
pe and to the different pathlengths  
Typically  $< 3 \text{ ns}$ ).

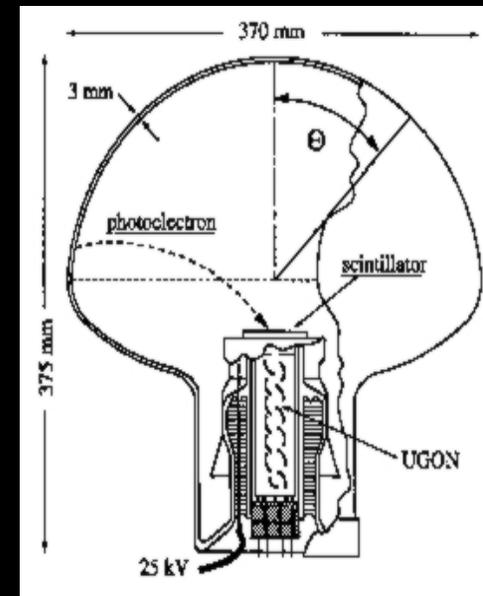


# Photosensors for Neutrino Telescopes



Time response: depends on electrode structure and supply voltage  
 Linearity: linear response of anode current vs incident light intensity

Smart ideas:  
 Baikal QUASAR



Short wave length limit  
 Determined by window material

Long wave length limits: determined by photocathode material  
 $h\nu$  too small to extract pe

# Parameters for selecting PMTs

Incident light conditions	
Light wavelength	Window material Photocathode spectral response
Light intensity	Number dynodes Dynode type Voltage applied to dynodes
Light beam size	Effective diameter (size) Viewing configuration
Speed of optical phenomenon	Time response

To which follows the selection of the circuit conditions: signal processing method (analog or digital method) and of the bandwidth of associated circuit

# New applications

Main drawbacks of PMTs are the bulky shape, the high price and the sensitivity to magnetic fields.

Photodiodes are semiconductor light sensors that generate a current or voltage when light illuminates the p-n junction. Allow detection of light in 200-1150 nm)

PIN photodiodes are Near IR detectors (InGaAs or Si) have excellent linearity with light, high speed, low noise and dark current but major drawback: low gain.

**Avalanche Si photodiodes (APDs)** have internal gain so are more sensitive and have much higher QE > 70%.

A = absorption region where there is an electric field that separates the generated electron-hole pairs and sweeps one carrier towards M = multiplication region. Here there is a high field that guarantees a high gain through impact ionization

G = 100 in Si APDs, 10-40 in Ge or InGaAs

G is limited by statistical fluctuations of avalanche multiplication

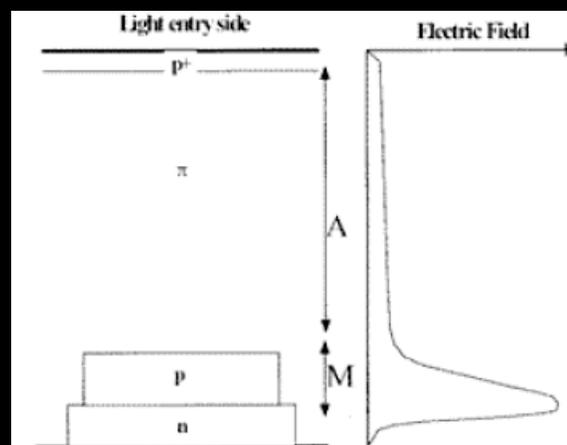


Figure 1: Reach-through APD Structure (Not to Scale)

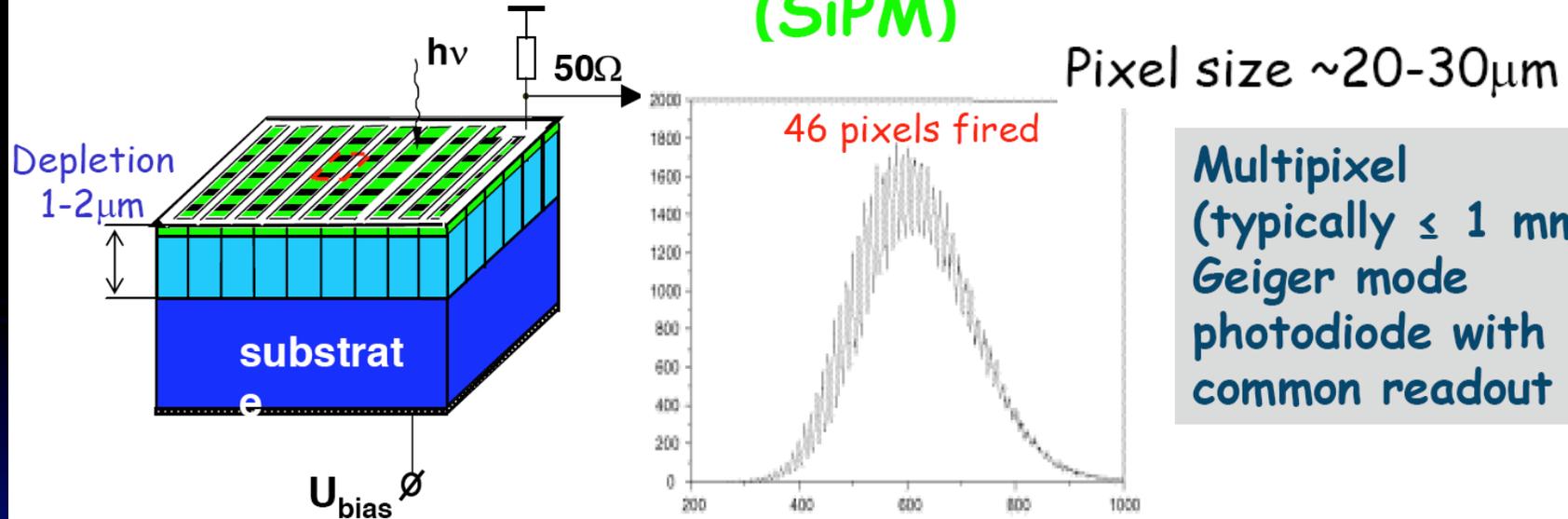
# Avalanche PhotoDiodes

- Operate a specially designed photodiode in reverse high voltage; a photon can induce a avalanche cascade of electron hole creation which is amplified in the strong field.
- Advantages of Avalanche photodiodes
  - very high quantum efficiency of more than 90% possible
  - very good time resolution:  $< 1$  nsec
  - high bandwidth
  - excellent photon counting resolution due to very large amplification (PMT limited by poisson statistics on the first dynode gain)
- Disadvantages or limitations:
  - Size: typically a few, up to 5 mm in diameter
  - Read the fine print of operational characteristics

# Silicon PMTs (Dolgoshein et al)

- ❑ **FIRST STEP:** SINGLE PHOTON AVALANCHE DIODE (SPAD), based on single pixel → “photon counter”
- ❑ **SECOND STEP:** from SPAD to

## Silicon Photomultiplier (SiPM)

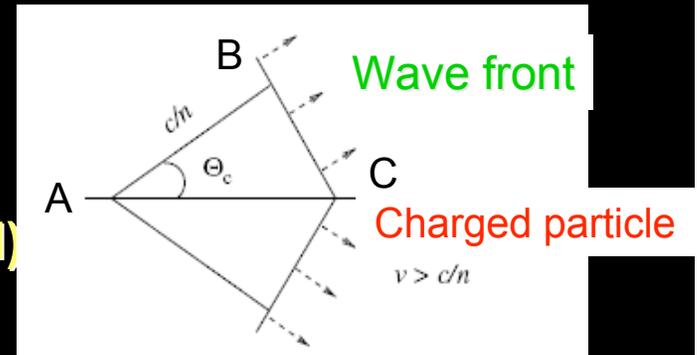


Multipixel  
(typically  $\leq 1 \text{ mm}^2$ )  
Geiger mode  
photodiode with  
common readout

Main problem inter-pixel cross-talk  
Fast Geiger discharge development  $< 500 \text{ ps}$   
Pixel recovery time  $\sim 20 \text{ ns}$

# The Cherenkov effect

A charged particle traveling in a dielectric medium with  $n > 1$  radiates **Cherenkov radiation** if its velocity is larger than the phase velocity of light  $v > c/n$  or  $\beta > 1/n$  (threshold)



The emission is due to an asymmetric polarization of the medium in front and at the rear of the particle, giving rise to a varying electric dipole moment.

Some of the particle energy is converted into light. A coherent wave front is generated moving at velocity  $v$  at an angle  $\theta_c$

If the media is transparent the Cherenkov light can be detected.

If the particle is ultra-relativistic  $\beta \sim 1$   $\theta_c = \text{const}$  and has max value

$$\cos \theta_c = \frac{AB}{AC} = \frac{\frac{c}{n}t}{\beta ct} = \frac{1}{\beta n}$$

In water  $\theta_c = 43^\circ$ , in ice  $41^\circ$

# The Cherenkov effect

The intensity of the Cherenkov radiation (number of photons per unit length of particle path and per unit of wave length)

$$\frac{d^2N}{dx d\lambda} = \frac{4\pi^2 z^2 e^2}{hc\lambda^2} \left(1 - \frac{1}{n^2\beta^2}\right) = \frac{2\pi z^2}{\lambda^2} \alpha \sin^2 \Theta_c$$

$$\alpha = \frac{2\pi e^2}{hc}$$

Number of photons/L and radiation length depends on charge and velocity of particle

Using light detectors (photomultipliers) sensitive in 400-700 nm for an ideally 100% efficient detector in the visible

$$\frac{dN_\gamma}{dx} = \int_{\lambda_1}^{\lambda_2} d\lambda \frac{d^2N_\gamma}{dx d\lambda} = 2\pi z^2 \alpha \sin^2 \Theta_c \int_{\lambda_1}^{\lambda_2} \frac{d\lambda}{\lambda^2} = 2\pi z^2 \alpha \sin^2 \Theta_c \left( \frac{1}{\lambda_1^2} - \frac{1}{\lambda_2^2} \right) = 490 z^2 \sin^2 \Theta_c \quad \text{photons/cm}$$

$$\frac{d^2N}{dx dE} = \frac{d^2N}{dx d\lambda} \frac{d\lambda}{dE} = \frac{\lambda^2}{2\pi hc} \frac{d^2N}{dx d\lambda}$$

$$E = h\nu = \frac{hc}{\lambda} = \frac{2\pi\hbar c}{\lambda}$$

$$\frac{d^2N}{dE dx} = \frac{\alpha z^2}{\hbar c} \sin^2 \theta_c = \frac{\alpha^2 z^2}{r_e m_e c^2} \left(1 - \frac{1}{\beta^2 n^2(E)}\right)$$

$$\approx 370 \sin^2 \theta_c(E) \text{ eV}^{-1} \text{ cm}^{-1} \quad (z = 1),$$

Energy loss is about  $10^4$  less than 2 MeV/cm in water from ionization but directional effect

# Example of radiators

Medium	$n-1$	$\gamma_{th}$	Photons/m
He (STP)	$3.5 \cdot 10^{-5}$	120	3
CO <sub>2</sub> (STP)	$4.1 \cdot 10^{-4}$	35	40
Silica aerogel	0.025-0.075	4.6-2.7	2400-6600
water	0.33	1.52	21300
Glass	0.46-0.75	1.37-1.22	26100-33100

# PID: Cherenkov detectors

Threshold Cherenkov detectors (common particles can be distinguished up to  $\sim 30$  GeV/c):

The Cherenkov threshold can be used to separate particles of momentum  $p$  and masses  $m_1$  and  $m_2 > m_1$ . The radiator medium can be chosen such that the heavier particle is just below threshold:  $\beta_2 \approx 1/n$

$$\sin^2 \Theta_c = 1 - \frac{1}{\beta_1^2 n^2} \approx 1 - \frac{\beta_2^2}{\beta_1^2} \approx 1 - \frac{\beta_2^2}{\beta_1^2} = 1 - \frac{E_1^2}{E_2^2} \approx \frac{c^2(m_2^2 - m_1^2)}{p^2}$$

$p \gg m_2$

Hence for a PMT with QE = 20%

$$N / L = 100 \sin^2 \Theta_c = 100 \frac{c^2(m_2^2 - m_1^2)}{p^2}$$

And to detect 10 photoelectrons one needs a radiator of length

$$L = 10 \frac{p^2}{c^2(m_2^2 - m_1^2)} \quad \text{in cm}$$

Eg for K/ $\pi$  separation at 1 GeV  $N_{pe}/L \approx 16/\text{cm}$  for  $\pi$  and by design 0 for K

# PID: Cherenkov detectors

## Differential or focusing Cherenkov detectors:

Angle of emission is measured with mirrors. There are 2 variations of the method:

- 1) If particles travel in the same direction the cone can be focused on to a slit diaphragm arranged to transmit light to a PMT. Scanning over a velocity range can be done adjusting the diaphragm to select different angles or modifying the refractive index by adjusting the pressure or composition of the gas
- 2) If particles do not travel along the same axis, the radiator can be contained between 2 spherical surfaces of radii  $R$  and  $2R$  centered on the target or interaction region. The outer surface is lined with a mirror which focuses the Cherenkov radiation into a ring at the inner detector surface. The radius of the ring depends on  $q$  and hence on the particle velocity. An electronic image of the ring is constructed eg passing photons through a suitable gas, and drifting the liberated photoelectrons on the wires of a planar MWPC or drift chamber (ring image Cherenkov detector)